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Stoichiometric Calculations are Based on Chemical Formulas Let's learn some terms used in Stoichiometry first.

Formula Mass: It is the sum of the atomic weights of the various atoms present in the molecule of the substance. For example, we can calculate the formula mass of  $\text{Na}_2\text{S}$  as

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 $2(23) + 1(32) = 78$

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9 The Chemical Reaction Equation and  
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The process can be defined as one or a  
series of operations in which physical



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and chemical ... Calculations of unsteady state processes are somewhat more complicated than that of

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Stoichiometry / , s t o i k i ' o m i t r i / is the calculation of reactants and products in chemical reactions in chemistry.

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Stoichiometry is founded on the law of conservation of mass where the total mass of the reactants equals the total mass of the products, leading to the insight that the relations among quantities of reactants and products typically form a ratio of positive integers.

**Stoichiometry - Wikipedia**

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