

Modern Chemistry Chapter 7 2 Review Answers

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Modern Chemistry Chapter 7 2

Modern Chemistry • CHAPTER 7 HOMEWORK 7-2 (pp. 206-211) VOCABULARY Match the symbol with the name that describes it. 1. Br- ____ a. nitrate 2. OH- ____ b. chlorite 3. NO 2 - ____ c. hydroxide 4. NO 3 - ____ d. bromide 5. O 2 - ____ e. nitrite 6. CN - ____ f. ammonium 7. ClO 2 - ____ g. oxide 8. NH 4 + ____ h. cyanide GRAPHIC ORGANIZER

HOMEWORK 7-2

54 CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS MODERN CHEMISTRY ... CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a.

7 Chemical Formulas and Chemical Compounds

HOMEWORK 7-2 Modern Chemistry Chapter 7. From the modern Chemistry Text Book made by HMH. STUDY. PLAY. What part of a chemical formula indicates how many atoms of each element are present in the compound? Subscript. The smallest unit of a substance that keeps all of the physical and chemical properties of that substance.

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HOMEWORK 7-1

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

CH 1 Reading Assignment Modern Chemistry. CH 1 Vocabulary-New. CH 1 Mixed Questions. CH 1 Matter & Energy Vocabulary-New. CH 1 Classifying Matter-New. CH 1 Classification of Matter ... Notes: Chapter 2 CH 2 Measurement & Calculations Honors Chapter 2. Reference: Sig Fig Rules. Metric Units Conversion. Review Activities: CH 2 Vocabulary Review ...

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-6 SKILL BUILDER Determine the oxidation numbers of these compounds and ions. N2 N3-N2O3 KNO 3 NH 4NO 3 K2Cr2O7 CO 3 2- NH 4 1+ ANOTHER SKILL BUILDER Name each compound according to the prefix system. Carbon(IV) iodide Sulfur(VI) oxide Arsenic(III) sulfide Nitrogen(III) chloride

HOMEWORK 7-6

7. Use the concept of electron configurations to explain why the number of valence electrons in metals tends to be less than the number in most nonmetals. Most metals have their outer electrons in s orbitals, while nonmetals have their outer electrons in p orbitals. MODERN CHEMISTRY CHEMICAL BONDING 47 Copyright © by Holt, Rinehart and Winston.