Ionic Diffusion In Membranes Rd Springer

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Ionic Diffusion In Membranes Rd

To diffuse through a membrane an ion must become associated with, or dissolved into, at least one component of that membrane. Diffusion is produced by thermal jumps from one molecular site to another. It is assumed that the electric field can change the binding properties between ions and membrane molecules.

Ionic diffusion in membranes | SpringerLink

A model for ion diffusion coefficients in ion exchange membranes, incorporating ideas from counterion condensation theory, was proposed to interpret the experimental results. The model predicted co-ion diffusion coefficients reasonably well with no adjustable parameters, while a single adjustable parameter was required to accurately describe counterion diffusion coefficients.

Ion Diffusion Coefficients in Ion Exchange Membranes ...

With this technique, the ionic composition of the cytosol is assumed to be strongly influenced (as result of diffusion) by the ionic composition of the solution contained in the electrode. If this assumption is valid for isolated cardiac myocytes, the technique would be particularly powerful for studying the dependence of their Na,K-pump on the ...

Ionic diffusion in voltage-clamped isolated cardiac ...

Effects of noradrenaline on the membrane potential and ionic permeability of parenchymal cells in the liver of the guinea-pig. Nature. 1969 Oct 4; 224 (5214):80–81. HECKMANN KD, PARSONS DS. Changes in the water and electrolyte content of rat-liver slices in vitro. Biochim Biophys Acta. 1959 Nov; 36:203–213. HODGKIN AL, KEYNES RD.

Ionic fluxes and permeabilities of cell membranes in rat liver

Hence, the facilitated transport constitutes a combination of physical diffusion enhanced by reversible chemical reaction and diffusion of the complex. In contrast the enhancement of physical absorption and diffusion in a novel catalytically active supported ionic liquid membrane (SILM) by simultaneous conversion of dissolved olefins to a more ...

Novel supported ionic liquid membranes for simultaneous ...

A Redox Pump for the Biological Performance of Osmotic Work, and Its Relation to the Kinetics of Free Ion Diffusion Across Membranes. 1953,,, 419-445. DOI: 10.1016/S0074-7696(08)61040-2. Tarsten Teorell. Transport Processes and Electrical Phenomena in Ionic Membranes. Progress in Biophysics and Biophysical Chemistry 1953, 3, 305-369. DOI: 10 ...

Diffusion and Membrane Permeability. \mid The Journal of ...

The variation in the ionic core volume and number density with EW can be explained on the basis of the membrane water content: in transversing the series BAM @ 735, 542 and 509 g mol -1 EW, the sulfonic acid content of the dry polymer increases but the membranes do not imbibe as much water as the lower EW membranes.

Ionic conductivity of proton exchange membranes ...

This article gives selected examples on the utilization of multispectral imaging for SpECM, a unique tool to study ISMs. SpECM has been used for the determination of the diffusion coefficients of different membrane ingredients and for the assessment of ionic impurities in ISE membranes and membrane plasticizers.

Multispectral imaging of ion transport in neutral carrier ...

A neat poly(vinylidene fluoride-co-hexafluoropropylene) membrane is impermeable to both the representatives of aliphatic hydrocarbons and branched hydrocarbons, namely hexane and isooctane, whereas the permeation flux is enhanced by the presence of 80 mass % of the ionic liquid 1-ethyl-3-methylimidazolium bis(trifluoromethylsulphonyl)imide in the membrane, as detailed in this work. The permeabilities of hydrocarbon vapours were determined from the binary mixture containing hydrocarbon and ...

Vapour permeation and sorption in fluoropolymer gel ...

lon-exchange membranes are traditionally used in electrodialysis or diffusion dialysis by means of an electrical potential or concentration gradient, respectively, to selectively transport cationic and anionic species.

Ion-exchange membrane - Wikipedia

For membranes as well as for aqueous chloride solutions, cation diffusion coefficients increased in the following sequence: Li + < Na + < Cs +. Cation and water molecule diffusion activation energies in temperature range from 20 °C to 80 °C were close to each other (about 20 kJ/mol).

Membranes | Free Full-Text | Hydration and Diffusion of H+ ...

The diffusion coefficient (D) of Ru (NH 3) 63+ through the membrane was found to be 2.2 (± 0.9) \times 10 -6 cm -2 s -1 from the measured flux, with the areal density, pore diameter, thickness, and...

Aligned Multiwalled Carbon Nanotube Membranes | Science

Important ions cannot pass through membranes by passive diffusion; if they could, maintaining specific concentrations of ions would be impossible. Osmotic pressure is directly proportional to the number of solute atoms or molecules; ions exert more pressure per unit mass than do non-electrolytes. Electrolyte ions require facilitated diffusion and active transport to cross the semi-permeable membranes.

41.1B: Transport of Electrolytes across Cell Membranes ...

Which function of cellular membranes is critical to the ionic and molecular organization of cells? The controlled transport of ions and molecules across the membrane Movement of a substance from an area of low concentration to an area of high concentration using energy obtained from ATP is called _____.

MindTap Chapter 5 Flashcards | Quizlet

Diffusion coefficients D 2 and D 3 (10 - 10 - 10 - 9 m 2/s) are typical of water molecules (or hydrated cation H + in acid ionic form of MSC membrane). The existence of water molecules with different translation mobility denotes on membrane heterogeneity.

Membranes | Free Full-Text | Hydration and Diffusion of H+ ...

Non-Ionic Diffusion of Some Organic Bases Through Cellular Membranes M. Kampp Medical Department B and the Central Laboratories of Clinical Biochemistry and Clinical Physiology, Bispebjerg Hospital, Copenhagen, Denmark & N. A. Lassen Medical Department B and the Central Laboratories

of Clinical Biochemistry and Clinical Physiology, Bispebjerg ...

Non-Ionic Diffusion of Some Organic Bases Through Cellular ...

Charge transport in 1-hexyl-3-methylimidazolium hexafluorophosphate ionic liquid in oxidized nanoporous silicon membranes is investigated in a wide frequency and temperature range by a combination of Broadband Dielectric Spectroscopy (BDS) and Pulsed Field Gradient Nuclear Magnetic Resonance (PFG NMR). By applying the Einstein-Smoluchowski relations to the dielectric spectra, diffusion ...

Charge transport and diffusion of ionic liquids in ...

The Goldman-Hodgkin-Katz voltage equation, more commonly known as the Goldman equation, is used in cell membrane physiology to determine the reversal potential across a cell's membrane, taking into account all of the ions that are permeant through that membrane. The discoverers of this are David E. Goldman of Columbia University, and the English Nobel laureates Alan Lloyd Hodgkin and ...

Goldman equation - Wikipedia

The present invention relates to a method for manufacturing a fuel cell electrode (E) by depositing a catalytic layer (2) on a diffusion layer (3), characterized in that said catalyst is deposited on the diffusion layer (3) by ionized physical vapor deposition (IPVD) in a vacuum chamber. The invention also relates to a method for manufacturing a fuel cell half-core comprising an ionic membrane ...

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