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Solubility and Temperature Gizmo :
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create a solution, observe that the chemical dissolves in the water at first, and then measure the concentration of the solution at the saturation point.

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water to create a solution, observe that the chemical dissolves in the water at first, and then measure the concentration of the solution at the saturation point.

Solubility and Temperature - ExploreLearning

The solubility of a given solute in a given

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solvent typically depends on temperature. Many salts show a large increase in solubility with temperature. Some solutes exhibit solubility that is fairly independent of temperature. A few, such as cerium (III) sulfate, become less soluble in water as temperature increases.

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Answer Key

Factors Affecting Solubility | Boundless Chemistry

112.45.12: The student knows the factors that influence the solubility of solutes in a solvent. 112.45.12.A: demonstrate and explain effects of temperature and the nature of solid solutes on the solubility of solids; Solubility and Temperature.

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112.45.12.B: develop general rules for solubility through investigations with aqueous solutions;

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the chemical dissolves in the water at first, and then measure the concentration of the solution at the saturation point.

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SPS6.c: Analyze and interpret data from a solubility curve to determine the effect

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of temperature on solubility. Solubility and Temperature. SPS7: Obtain, evaluate, and communicate information to explain transformations and flow of energy within a system. SPS7.a: Construct explanations for energy transformations within a system.

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Answer Key **Science Simulations**

The Solubility and Temperature Gizmo allows students to measure the effect of temperature on solubility of two different substances, including salt, by adding varying amounts of a chemical to a beaker of water to create a solution and then measuring the concentration of the solution at the saturation point.

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Reporting System are protected by US Patent No. 10,410,534. 110 Avon Street, Charlottesville, VA 22902, USA

ExploreLearning Gizmos: Math & Science Simulations

Solubility and Temperature C.10.E: The student is expected to: distinguish among types of solutions such as

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electrolytes and nonelectrolytes;
unsaturated, saturated, and
supersaturated solutions; and strong
and weak acids and bases;

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the solution at the saturation point.

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solubility, magnetic properties, melting and boiling points, and know that these properties are independent of the amount of the sample. Explain whether the solubility of a substance is

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determined by the solute, the solvent or both. Solubility and Temperature
SC.8.P.8.4 – Classify and compare substances on the basis of characteristic

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gizmo explorelearning add varying amounts of a chemical to a beaker of water to create a solution observe that the chemical dissolves in the water at first and then measure the concentration of the solution at the saturation point

Solubility And Temperature Gizmo Answers

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In the Solubility and Temperature Gizmo™, you will study how temperature affects how much solute will dissolve in a solution. To begin, check that Potassium nitrate is selected and the Temp. of the water is 20 °C. Click OK.

1. In this solution, what is the solute? _____ What is the solvent? _____

2.

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