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called binary solution. Solute and Solvent

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1. All compounds present, both the reactants and products after the double replacement reaction are soluble in water because of the presence of a group 1 ion. All ions are present in solution because everything is soluble in water. ( $\text{Na}_2\text{CrO}_4 + \text{K}_2\text{CO}_3 \rightarrow \text{K}_2\text{CrO}_4 + \text{Na}_2\text{CO}_3$ )

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