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## **Chemical Quantities Review Sheet**

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Review Sheet Answers following calculations. Show all of your work, include your units and round answers to the correct number of significant figures. How many atoms are equal to 4.61 moles of carbon?  $2.78 \times 10^{24}$  atoms C.

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## Chemical

### Quantities Review

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Chemical Quantities -  
mutic.weebly.com The  
mass of a mole of a  
compound is equal to  
the total mass of all  
the elements of the  
compound. Each  
element has an atomic  
mass that can be found  
in the periodic table.

## **Chemical Quantities**

### **Section Review**

#### **Answers**

SCH3U Unit 3:  
Quantities in Chemical  
Reactions Review

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Sheet Important  
equations: 1. Calculate  
the following: a) How  
many moles are  
present in 8.5 g of  
magnesium nitrate,  
 $\text{Mg}(\text{NO}_3)_2$ ? b) How  
many atoms are there  
in 8.5 g of  $\text{Mg}(\text{NO}_3)_2$ ?  
21c) How many moles  
are there in  $4.20 \times 10$   
molecules of methane?

## **SCH3U Unit 3: Quantities in Chemical Reactions Review Sheet**

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**Answers**

\*Answer sheet DUE by  
END of class

\*ANSWERS NB. TYPO ...  
- Work on lab report OR  
Review Work for unit  
test UNIT TEST-

Quantities in Chemical  
Reactions: ... UNIT  
TEST: Quantities in  
Chemical Reactions ...

**UNIT 3- Quantities in  
Chemical Reactions -  
Ms. Gauthier**

Chemical Quantities  
Review Sheet PART 1:

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Short Answers

THE MOLE AND  
CONVERSION - A  
conversion factor  
allows for the  
conversion between  
units. - Representative  
particle - species  
present in a substance  
o Include atoms,  
molecules, ion, formula  
units - Conversion  
Factor 1: 1 mole =  $6.02 \times 10^{23}$  items o  
Avogadro's number =  
 $6.02 \times 10^{23}$  items

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## **Chemical Quantities Review Sheet**

Chemical Quantities

THE MOLE AND

QUANTIFYING MATTER

10.1 The Mole: A

Measurement of Matter

Essential

Understanding The

mole represents a

large number of very

small particles.

Reading Strategy

Fruyer Model The

Fruyer Model is a

vocabulary

development tool. The

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Sheet Answers

center of the

**Chemical Quantities  
- AP Biology**

Answer Key: Chemical  
Reactions Review.

reactants on the left,  
products on the right.

iodine, bromine,

chlorine, fluorine,

oxygen, nitrogen,

hydrogen. a whole

number in front of an

element or compound

in a chemical equation;

used to balance the

equation.

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Quantities Review

**Answer Key:**  
**Chemical Reactions**  
**Review**

Chemical Quantities  
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[EBOOK] Chapter 7  
Chemical Quantities  
Answer Key Prentice ...  
Chapter 10 Chemical  
Quantities Worksheet  
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to

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**Name Date Class  
period ...**

The atomic mass of an element is the mass of a single atom in atomic mass units. chapter 10

chemical quantities  
test. us Chapter 10 -

Chemical Quantities  
Pre-AP Chemistry

Chemical Quantities  
Review Sheet

Measurements and.  
Chapter 8: Quantities  
in Chemical Reactions



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DISCUSSION PLAN

Chapter Summary

Chapter 8 launches discussion about stoichiometry with comparing a chemical reaction to a cooking ...

**Chapter 10 Chemical Quantities Practice Problems Answers ...**

Chemical Quantities A.

Completion Section

Date Use this

completion exercise to check your knowledge of the terms and your

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understanding of the  
concepts introduced in  
this chapter. Each  
blank can be  
completed with a term,  
short phrase, or  
number. 6-2 6-2 6-3  
6-2 6-3 6-3 6-3 6-3 6-3

**Baltimore**

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Chapter 10 Chemical

Quantities Practice

Problems Answers With

Work

**Chapter 10 Chemical**

*Page 18/24*

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## **Quantities Practice Problems Answers ...**

Chemical reactions relate quantities of reactants and products. Chemists use the mole unit to represent  $6.022 \times 10^{23}$  things, whether the things are atoms of elements or molecules of compounds. This number, called Avogadro's number, is important because this number of atoms or molecules has the

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Sheet Answers  
same mass in grams as  
one atom or molecule  
has in atomic mass  
units.

## **CH104: Chapter 6 - Quantities in Chemical Reactions**

...

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Chapter 6 Chemical  
Quantities study guide  
by owasley15 includes  
38 questions covering  
vocabulary, terms and

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## **Chapter 6 Chemical Quantities**

### **Flashcards | Quizlet**

Labs. Do the "Percent Sugar in Bubble Gum" lab and have students calculate the molar mass of the sugar, convert the mass of the sugar to moles, and determine the number of molecules of

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sugar in the gum. Do

"The Volume of 1 Mole  
of Hydrogen Gas" lab.

Or try "The

Determination of

Relative Atomic Mass"

lab. Do the "Molar

Volume of a Gas"

lab. Try the "Finding the

Formula of an Oxide of

Copper" lab.

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