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Collision Theory Gizmo Answer Key The Collision Theory Gizmo™ allows you to experiment with several factors that affect the rate at which reactants are transformed into products in a chemical reaction. You will need blue, green, and orange markers or colored pencils for the first part of this activity. 1.

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9/9/2019 Collision Theory Gizmo : ExploreLearning 2/5 A. Graph A B. Graph B C. Graph C D. Graph D Explanation: According to collision theory, chemical reactions occur when molecules collide at a certain orientation and with enough energy. When the temperature is raised, the average speed of the molecules rises as well. This increases the number of collisions between

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molecules, and the energy ...

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Collision Theory Observe a chemical reaction with and without a catalyst. Determine the effects of concentration, temperature, surface area, and catalysts on reaction rates. Reactant and product concentrations through time are recorded, and the speed of the simulation can be adjusted by the user.

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advertisement. Collision Theory Gizmo Rate of Reactions. A chemical reaction causes the chemical compositions of substances to change. Reactants are substances that enter into a reaction, and products are substances produced by the reaction. The Collision Theory Gizmo™ allows you to experiment with several factors that affect the rate at which reactants are transformed into products in a chemical reaction.

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Collision theory all reactions require activation energy, so reactions will only occur when atoms, and compounds collide with enough energy. How does increased concentration increase the rate of reaction?

## Collision Theory Flashcards | Quizlet

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When the two compounds collided, the blue compound took one of the compounds from the green compound and drifted away. When two reactant molecules meet, they form a temporary structure called an activated complex. The activated complex breaks up into the product molecules. 2.



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